

**Info about the test:**

- 1) It is **open book**/notes/our Moodle site/our Google Drive with solutions.
- 2) **Please study for it.** Incentivizing studying is the only good reason I can think of for giving a test!
- 3) You can schedule taking it in any 3 1/2 hour block of time that is convenient for you during the exam period from May 13 - May 20.
- 4) A Google Doc will be available soon, for you to specify the time you want to take the test. **Thank you for, before May 13, committing to a time that works for you.**
- 5) **I'll email the test to you at your start time.** You have 3 1/2 hours to email answers back to me. I'll do my best to write a test that, if one has studied, takes comfortably less than 3 hours to do. I'm adding an extra half hour for you to check work, scan, and be sure the scans you email back are legible.
- 6) If you have an extra time accommodation, that extra time will be added on.

-----Practice test starts here-----

**Instructions:**

- Please write as **legibly** as you can so that your scans that you email back to Amy are highly readable.
- There are 7 questions on this test. Please **choose 5 of them to do for credit.**
- Submit to me **only the 5 solutions** that you want me to evaluate.  
(To be fair to everyone, if you submit more I will only look at the first 5.)
- Please show all reasoning and work leading to your answer. **Credit will strongly emphasize your process and your work ...** the way you explain the problem and your journey toward the answer ... not just the answer itself.
- **You have 3 1/2 hours** to email your work back to Amy.
- **Good luck!!!**

**Useful data and conversion factors:**

Ideal gas constant  $R = 8.31 \text{ J/mol} \cdot \text{K}$ .

Boltzmann's constant:  $k = 1.38 \times 10^{-23} \text{ J/K}$

Planck's constant:  $h = 6.63 \times 10^{-34} \text{ J s}$

Proton mass:  $1.66 \times 10^{-27} \text{ kg}$

Electron mass:  $9.1 \times 10^{-31} \text{ kg}$

$1 \text{ eV} = 1.60 \times 10^{-19} \text{ Joule}$

Absolute zero:  $0\text{K} = -273.15 \text{ }^\circ\text{C}$

speed of light:  $c = 3.00 \times 10^8 \text{ m/s}$

**1. Near the triple point of a substance:**

- the liquid-gas coexistence curve has  $P(T) = Ae^{-B/T}$
- the solid-gas coexistence has  $P(T) = Ce^{-D/T}$

In the expressions above,  $A$ ,  $B$ ,  $C$  and  $D$  are constants

Please find the triple point temperature.

2. A system of  $N$  electrons (which have their usual mass and spin  $1/2$ ) exist in a nanowire of length  $L$ , which we can model as a one-dimensional box. For this problem, ignore interactions between electrons, treating them like a 1d ideal fermionic gas. What is the Fermi temperature,  $T_F$ , in degrees K, if  $L = 10^{-6} \text{ m}$  and  $N = 10^8$ ?

3. Photons are massless bosons with two polarization states and  $\mu = 0$ . For a box of volume  $V$  and temperature  $T$ , please derive the mean number of photons in the box,  $N$ .

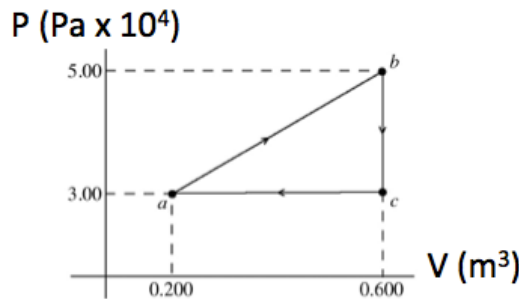
4. Consider a paramagnetic system of  $N$  spins that can take on, not two values, but *three* values:  $s_i = -1, 0, 1$ . The spins sit on a lattice. They do not interact with each other, but they do interact with a magnetic field so  $E_i = -\mu B s_i$  is the energy of the  $i^{\text{th}}$  spin. (Here  $\mu$  is a magnetic moment.) What are the free energy,  $F$ , and the average magnetization  $\langle M \rangle$ , of a system of these spins at temperature  $T$ ? (Hint: Does your  $\langle M \rangle$  come out to have reasonable limits for low and high  $T$ ?)

5. The function  $U(S, V, N)$  is a "master function": all thermodynamic quantities of interest can be obtained by taking suitable derivatives of  $U(S, V, N)$ . For the variables  $T, V, N$  the master function is the Helmholtz free energy  $F(T, V, N)$ . While energy  $U(T, V, N)$  is a valid function, it has lost its "master function" status. Demonstrate this by considering two different substances, A and B, with free energies:

$$F_A(T, V, N) \text{ and } F_B(T, V, N) = F_A(T, V, N) + aTV^2/N \text{ where } a \text{ is a constant.}$$

Show that these two substances have identical energies  $U(T, V, N)$  but different equations of state  $P(T, V, N)$ .

6. The  $PV$  diagram below is for an ideal gas, working in a cycle from  $a \rightarrow b \rightarrow c \rightarrow a$ .



Is the net heat,  $Q$ , absorbed by the gas positive, negative or zero? If nonzero, find the net heat,  $Q$ .

7. Two friends, Jay and Kay, challenge each other to a 1d random walk contest to see who will get further from their starting point at  $x=0$ . Jay takes steps to the right and left with an equal probability of  $1/2$ , and takes  $2N$  steps of size  $a$ . Kay takes steps to the right with probability  $3/4$ , and left with probability  $1/4$ , and takes  $N$  steps of size  $a$ .

For each person, Jay and Kay, please calculate

a) Their mean displacement  $\langle x \rangle$  from the origin

b) Their *rms* displacement  $\sqrt{\langle x^2 - \langle x \rangle^2 \rangle}$  from the origin

c) Comment on which quantity is bigger for which friend (or if both are the same?) and do a reality-check of why this makes sense to you.

**Answers:**

1.  $T_{\text{triple}} = \frac{B - D}{\ln A - \ln C}$

2.  $T_F = 1.1 \times 10^3 \text{ K}$

(Some explanation: You need to use the 1d density of states for particles in a box, and integrate over all k states up to  $k_F$  and set this integral equal to N. You should get  $k_F = N \pi / 2L$  where the "2" in denom comes because there will be 2 electronic spin states occupying each k state. Then you use the relationship between  $k_F$  and  $T_F$  and plug in numbers.)

3. Your derivation hopefully leads to:  $N = 2.404 \frac{V(kT)^3}{\pi^2 c^3 \hbar^3}$

4.  $F = -NkT \ln[1 + 2 \cosh(\mu B/kT)]$ ,  $\langle M \rangle = 2 N \mu \sinh(\mu B/kT) / [1 + 2 \cosh(\mu B/kT)]$ .

5. Both  $U_A$  and  $U_B$  come out to be  $F_A - T \left( \frac{\partial F_A}{\partial T} \right)_{V,N}$ . This comes from the definition

$U = F + TS$  and obtaining  $S$  as a partial derivative of  $F(T, V, N)$ . If you look for  $P(T, V, N)$  in a similar way as the appropriate partial derivative of  $F$ , with the appropriate quantities held constant, you find  $P_B = P_A - 2aVT/N$ .

6. Positive heat is absorbed (thus, positive work is done by the gas on the outside world).

$Q = 4000 \text{ J}$

7.

a)  $\langle x \rangle$  for Jay is 0 while for Kay it is  $\frac{N}{2}a$

b)  $x_{rms}$  for Jay is  $\sqrt{2Na}$  while for Kay it is  $\sqrt{\frac{3}{4}Na}$ .

c) Jay does an "unbiased" walk with  $p = q$ . So by symmetry, their mean displacement is zero. Kay's walk is biased to the right so it makes sense they move to the right, with nonzero mean displacement of  $(p-q)Na$ , given  $N$  is the number of steps of size  $a$  that Kay takes. On the other hand,  $x_{rms}$  is proportional to the width of the underlying binomial distribution. This is proportional to  $pq$  and to the number of trials. Here,  $pq$  is larger for Jay ( $1/4$  vs  $3/16$ ) and also Jay does a larger number of trials ( $2N$  for Jay vs.  $N$  for Kay). Thus, while Kay moves further as measured by the mean displacement, Jay moves further as measured by the *rms* displacement.