My observations about reading this week ...

0. Department of redundancy department:

- All texts talk about the van der Waals gas and its phase transition.
- All texts talk about the **Maxwell construction**. This permits one to follow the equilibrium liquid gas coexistence curve, rather than follow a spurious route in (P,V,T) space where the Gibbs free energy is not a minimum. This produces what some people call the "**lever rule**" ... the specific volumes of liquid and gas when they coexist. This equation is G&T (7.65), B&B (26.30), and Schroeder (5.57).
- Both B&B and G&T talk about the **virial series**. (So does Schroeder but I didn't assign that section.)

1. G&T probably has the best coverage this week:

Downside? As usual it is more lengthy than B&B and more mathematical than Schroeder. **Upsides:**

- It has a careful discussion of how to implement the calculation of the lever rule, and find the densities of liquid and gas when you are close to the critical point.
- It has a careful discussion of the Gibbs free energy as a function of density, reminiscent of the discussion of mean field theory for magnets. This feeds into an optional part of homework problem 5 where we see why the system "chooses" either liquid (high density) or gas (low density) or a coexistence.
- It alone has a discussion of critical exponents for the van der Waals model.
- **2. B&B** has a few **unique** and/or **good** things:
- Section 28.6 on Colligative properties is necessary reading this week. The altering of a phase transition temperature when solute is present is a key idea:-).
- Box on p. 298 has Z(T,V,N) and F for the van der Waals gas.
- Section 26.4 has a concise treatment of the Law of Corresponding States, with a graph.
- **Section 28.1** shows S vs. T with the jump in S indicating L, a latent heat.
- **P. 329** discusses **metastability**. (The ensuing discussion of nucleation and droplets is great but not required.)
- Section 28.5 introduces the Gibbs phase rule.
- Section 28.7 defines types of phase transitions by their order, as well as the type of symmetry which is broken.
- **3.** There is **nothing unique in Schroeder 5.3** that can't be found in other books. Still, he's super readable as he explains:
 - The boundary between two phases
 - The Clausius-Clapeyron (CC) Eq.
 - The van der Waals model
 - How to navigate the Maxwell construction.

Schroeder has some awesome problems. Thus, you will need to look at Schroeder to find some of the homework, and perhaps look near the problem to read some useful numbers or view a figure. Yet all the learning you need to do these problems can be obtained from other readings.

Schroeder 8.1 is not assigned, even though it discusses the virial series, which is a topic this week. It dives into a discussion of a diagrammatic expansion for which we don't have time. I am

assigning Schroeder Problem 8.10 , but it's totally doable from G&T's treatment of the Lennard-Jones model.