

The background features abstract, overlapping green geometric shapes, primarily triangles and polygons, in various shades of green, creating a modern and dynamic visual effect.

Lattice Gas

Ella Small

- Presentation: The lattice gas Please do G&T Problem 5.23. Explain to us how a lattice gas is like or unlike an Ising model. Is there a critical temperature T_c ? What happens to the gas when the temperature crosses this value? What happens when you lower the temperature very quickly, a “quench”? What happens when you turn on a gravitational field?

Lattice gas

- ▶ Is essentially an Ising model in 2D since in the fluid model $s=+1$ correlates to a particle at the grid point and $s=-1$ means no particle is present
 - ▶ The difference between the two is a change of variable from s (spin) to n (occupancy)

$$n_i \equiv (s_i + 1)/2,$$

- ▶ Only one particle per site = hard core repulsion
 - ▶ Two things cannot occupy the same space at the same time

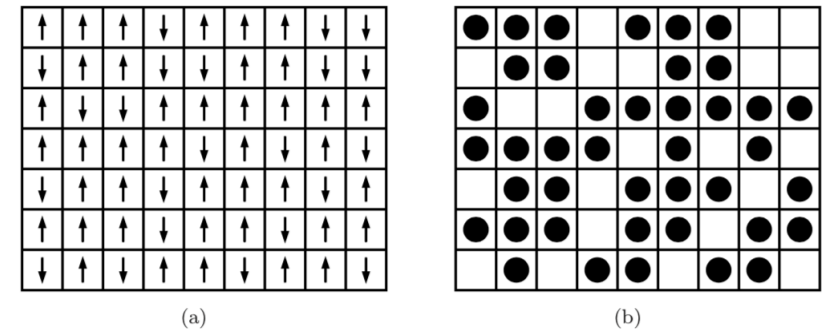


Figure 5.17: (a) A typical microstate of the Ising model. (b) The same microstate in the lattice gas picture with spin up replaced by a particle and spin down replaced by an empty cell.

(a) What is the value of the critical temperature T_c for the lattice gas in two dimensions (in units of u_0/k)?

- We can use the derivation the textbook uses for the Onsager solution since it was derived for a 2D Ising model

$$\frac{kT_c}{J} = \frac{2}{\ln(1 + \sqrt{2})} \approx 2.269.$$

- Except in this case

$$E = -u_0 \sum_{i,j=\text{nn}(i)}^N n_i n_j - \mu \sum_{i=1}^N n_i + N(H - 2J).$$

- and

$$u_0 \equiv 4J$$

- Replacing J by $u_0/4$ we get

- $T_c = J/k * 2.269 = u_0/4k * 2.269 = 0.567 * u_0/k$

(b) Program `LatticeGas` simulates the lattice gas on a square lattice of linear dimension L . The initial state has all the particles at the bottom of the simulation cell. Choose $L = 32$ and set the gravitational field equal to zero. Do a simulation at $T = 0.4$ with $N = 600$ particles. After a few Monte Carlo steps you should see the bottom region of particles (green sites) develop a few small holes or bubbles and the unoccupied region contain a few isolated particles or small clusters of particles. This system represents a liquid (the predominately green region) in equilibrium with its vapor (the mostly white region). Record the energy. To speed up the simulation set `steps per display` equal to 100.

QuickTime Player File Edit View Window Help

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LatGas2DApp Controller

File Edit Display Help

Input Parameters

Name	Value
Length	32
Number of particles	600
Temperature	0.4
Gravitational field	0
steps per display	100

Initialize Step Reset Zero averages

Messages

clear

Memory in use: 11MB of 3641MB

ional Lattice Gas

interpreted to model a fluid instead of a magnetic
a value of 0 for an unoccupied site and 1 for a site
on of two nearest neighbor occupied sites is $-u_0$.

to simulate a system in equilibrium with a heat bath
the volume or number of sites is fixed. The trial
f its nearest neighbors and exchanging the values at
his trial change is called spin-exchange dynamics or
 T_c in this system is about 0.567 (one-quarter of the
sity corresponds to 50% occupancy.

A gravitational field can also be added to the simulation such that each particle has an
additional energy equal to gh , where g is the field and h is the vertical position of the particle. We
assume the particles have unit mass and that the temperature is given in terms of u_0 .

Problems

The goal of the following problems is to observe two phase coexistence and spinodal
decomposition.

1. Run the program at $T = 0.4$ with 600 particles on a 32×32 lattice in zero gravitational field. You
should see one region of particles (green sites) with a few small holes or bubbles and another
region with just a few isolated particles or small clusters of particles. This system represents a
liquid (the predominately green region) in equilibrium with its vapor (the mostly white region).
Record the energy. To speed up the simulation set steps per display to 100.

Simulations in Thermal Physics

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SIMULATION
EJS MODEL
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PROGRAMM
TOOLS
JS/HTML
MATERIALS
BROWSE MA
RELATED SI
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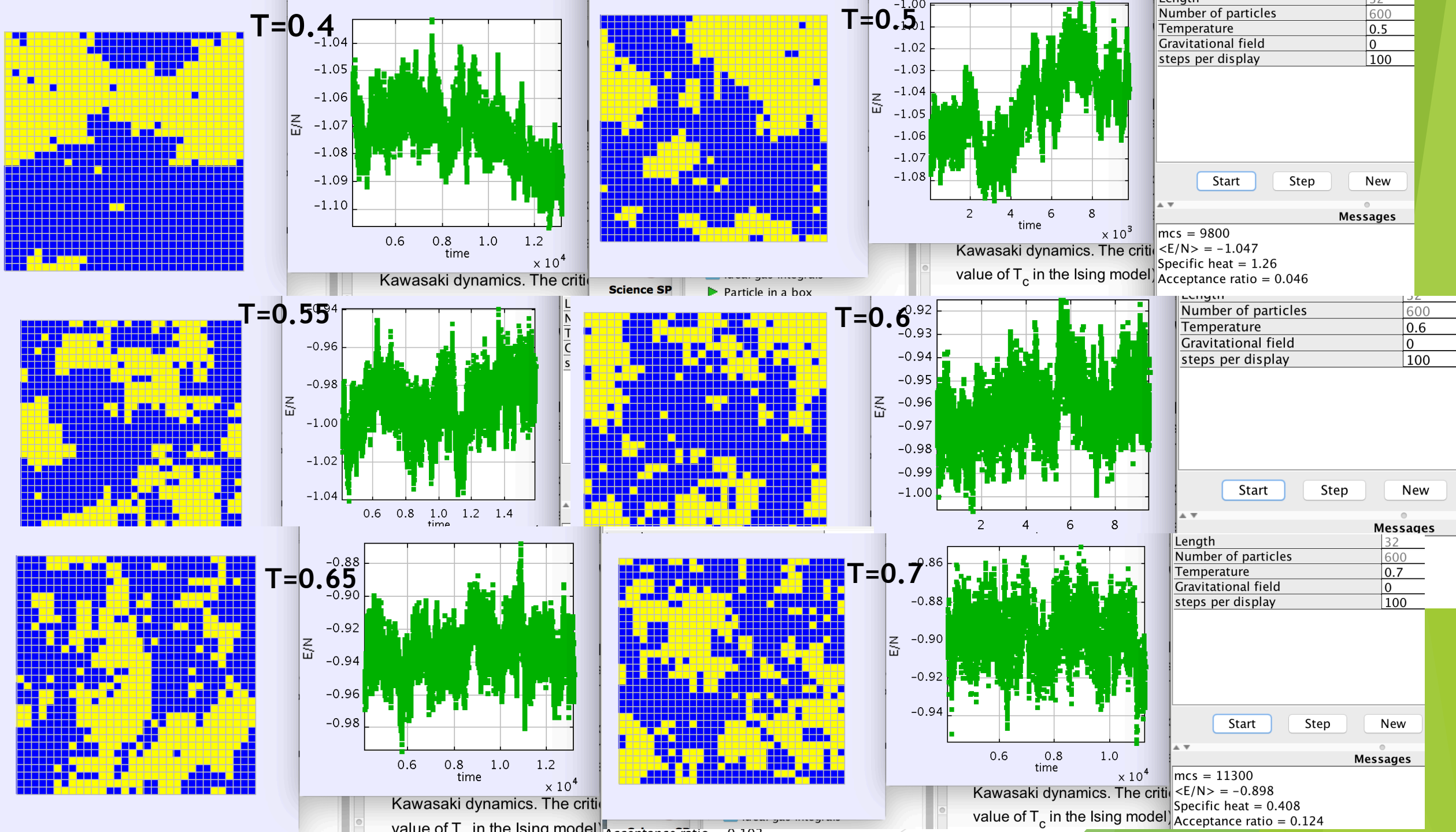
Science SP
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Random walks
Multiple coin toss
Binomial distribution
Central limit theorem
Monte Carlo estimation
Multiplicative processes
Boltzmann probability
Einstein solid: Temperatur
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Einstein solid: Entropy
Einstein solid: heat bath
Identification of the tempe
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Hard disks
Ising model
Ideal gas integrals
Particle in a box
Second virial coefficient
Diffusive equilibrium
Chemical demon
Estimating the chemical potential
Potts model
XY model
FPU problem
Ideal quantum gases
Percolation
Renormalization group (percolat
Lattice gas
Diffusion on a lattice

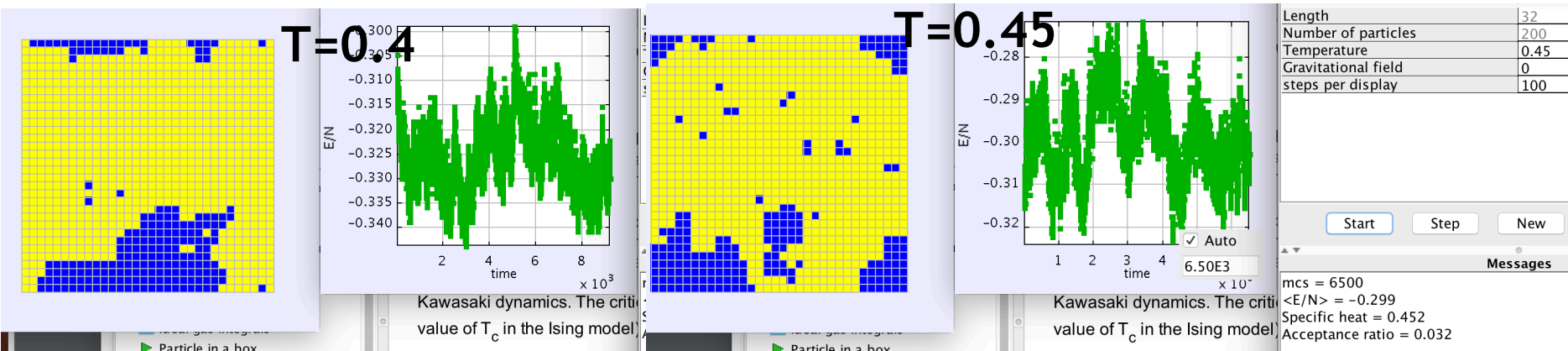
- (c) Increase the temperature in steps of 0.05 until $T = 0.7$. At each temperature run for at least 10,000 mcs to reach equilibrium and then press the **Zero Averages** button. Run for at least 20,000 mcs before recording your estimate of the energy. Describe the visual appearance of the positions of the particle and empty sites at each temperature. At what temperature does the one large liquid region break up into many pieces, such that there is no longer a sharp distinction between the liquid and vapor region? At this temperature there is a single fluid phase. Is there any evidence from your estimates of the energy that a transition from a two-phase to a one-phase system has occurred? Repeat your simulations with $N = 200$.

Temperature	$\langle E/N \rangle$
0.4	-1.106
0.45	-1.074
0.5	-1.047
0.55	-0.996
0.6	-0.962
0.65	-0.934
0.7	-0.898

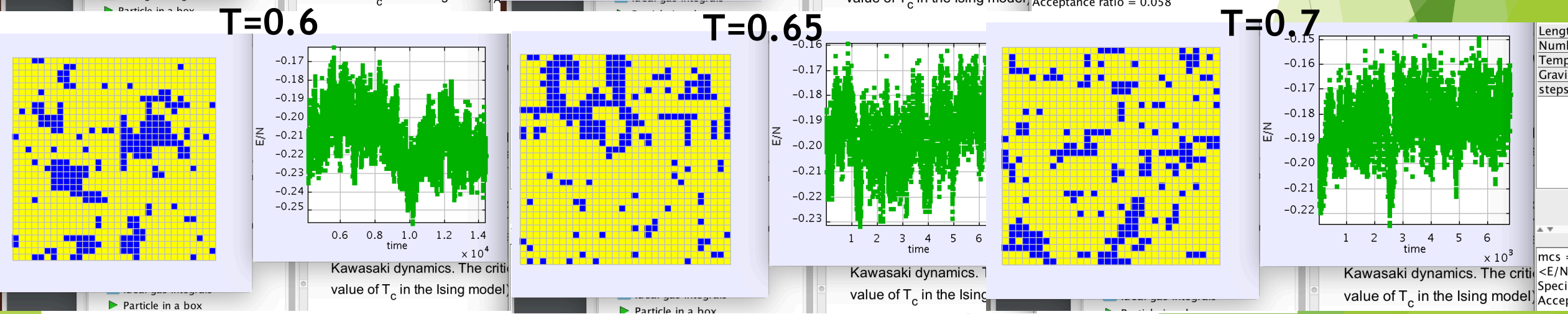
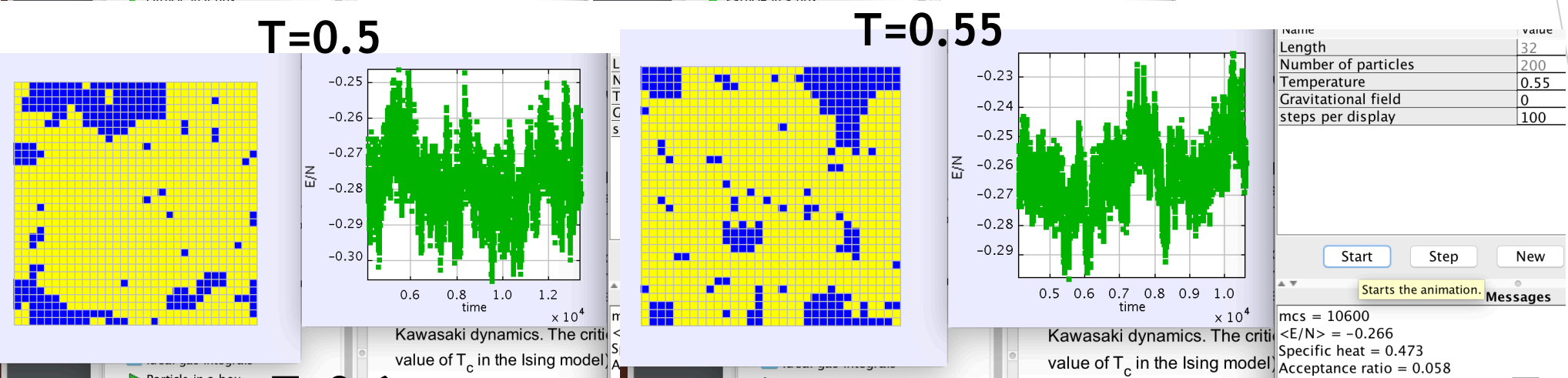


Repeating with N=200

Temperature	$\langle E/N \rangle$
0.4	-0.326
0.45	-0.299
0.5	-0.276
0.55	-0.266
0.6	-0.214
0.65	-0.196
0.7	-0.183



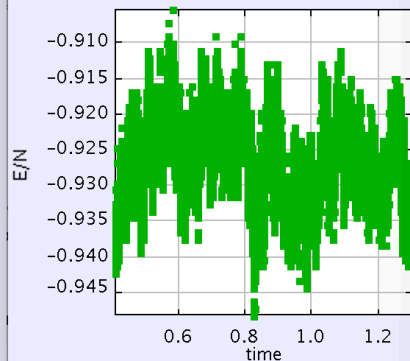
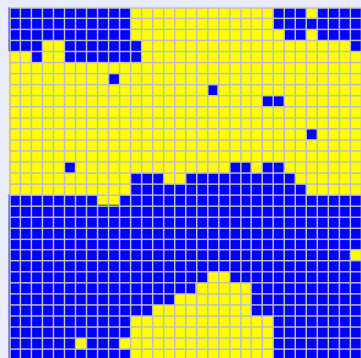
Similar results here as in the N=600 case



- (d) Repeat part (c) with $N = 512$. In this case the system will pass through a critical point. The change from a one phase to a two-phase system occurs continuously in the thermodynamic limit. Can you detect this change or does the system look similar to the case in part (c)?

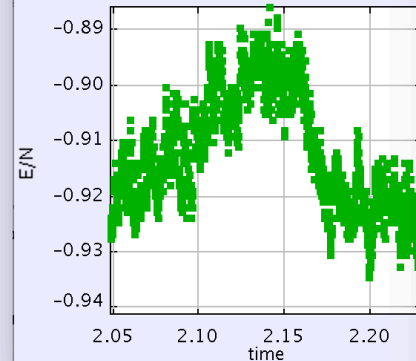
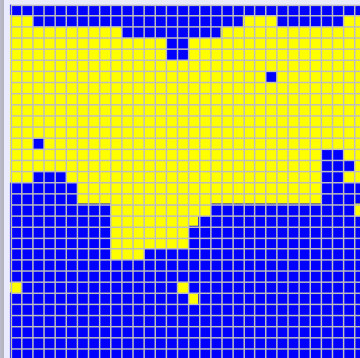
Temperature	$\langle E/N \rangle$
0.4	-0.928
0.45	-0.911
0.5	-0.887
0.55	-0.852
0.6	-0.789
0.65	-0.751
0.7	-0.722

T=0.4



Kawasaki dynamics. The critical value of T_c in the Ising model

T=0.45



Kawasaki dynamics. The critical value of T_c in the Ising model

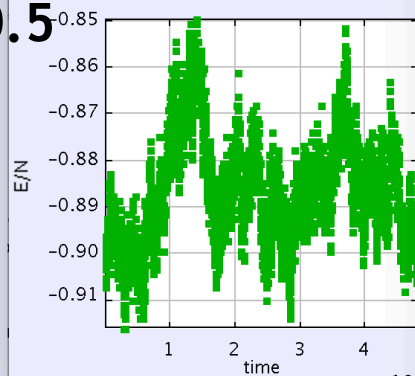
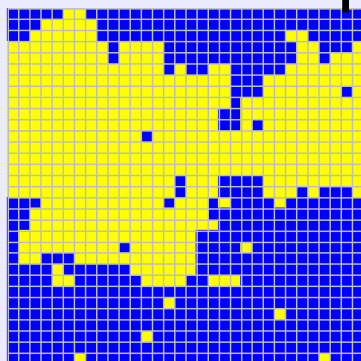
Length	32
Number of particles	512
Temperature	0.45
Gravitational field	0
steps per display	100

Start Step New

Messages

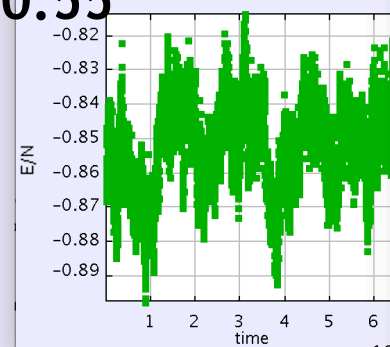
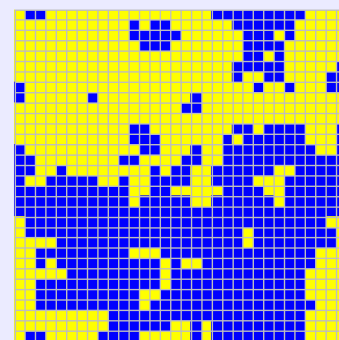
mcs = 22300
 $\langle E/N \rangle = -0.911$
 Specific heat = 0.473
 Acceptance ratio = 0.03

T=0.5



Kawasaki dynamics. The critical value of T_c in the Ising model

T=0.55



Kawasaki dynamics. The critical value of T_c in the Ising model

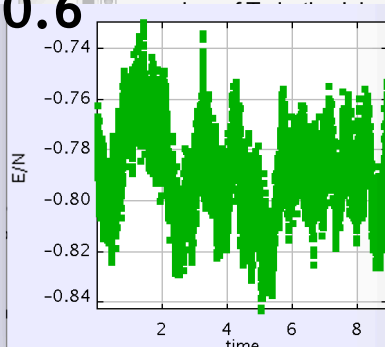
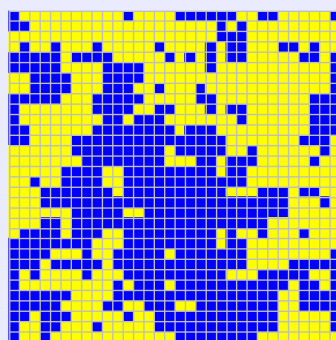
name	value
Length	32
Number of particles	512
Temperature	0.55
Gravitational field	0
steps per display	100

Start Step New

Messages

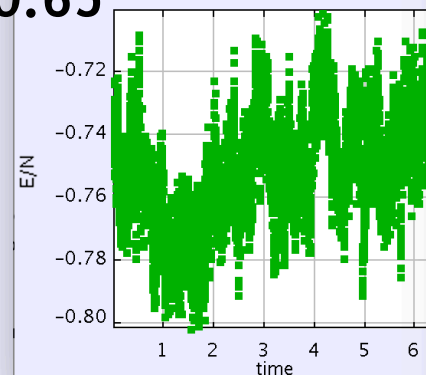
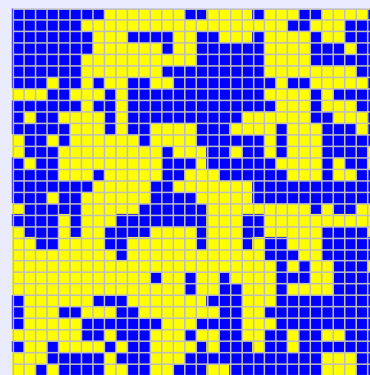
mcs = 6500
 $\langle E/N \rangle = -0.852$

T=0.6



Kawasaki dynamics. The critical value of T_c in the Ising model

T=0.65



Kawasaki dynamics. The critical value of T_c in the Ising model

Length	32
Number of particles	512
Temperature	0.65
Gravitational field	0
steps per display	100

Start Step New

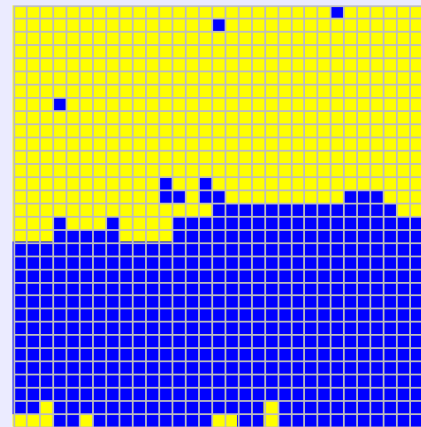
Messages

mcs = 6400
 $\langle E/N \rangle = -0.751$
 Specific heat = 0.811
 Acceptance ratio = 0.105

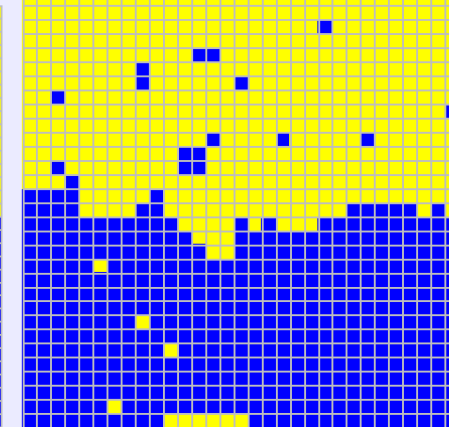
(e) If we include a gravitational field, the program removes the periodic boundary conditions in the vertical direction, and thus sites in the top and bottom rows have three neighbors instead of four. The field should help define the liquid and gas regions. Choose $g = 0.01$ and repeat the above simulations. Describe the differences you see.

Temperature	$\langle E/N \rangle$
0.4	-0.946
0.45	-0.888
0.5	-0.867
0.55	-0.836
0.6	-0.801
0.65	-0.73
0.7	-0.724

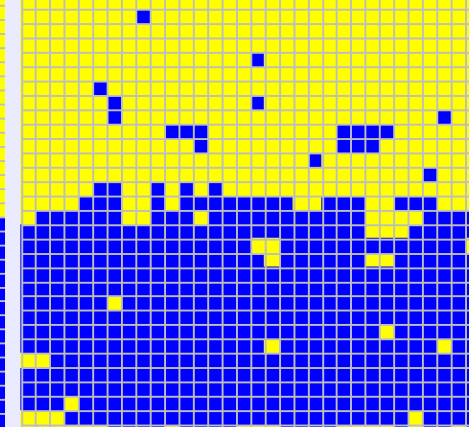
T=0.4



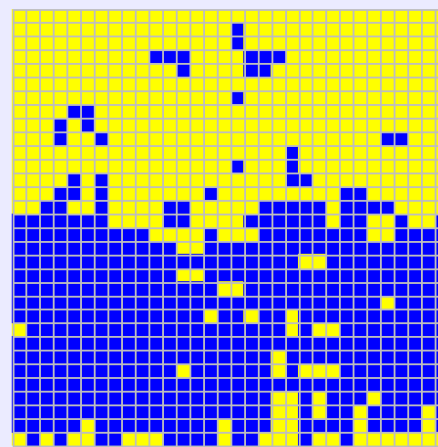
T=0.45



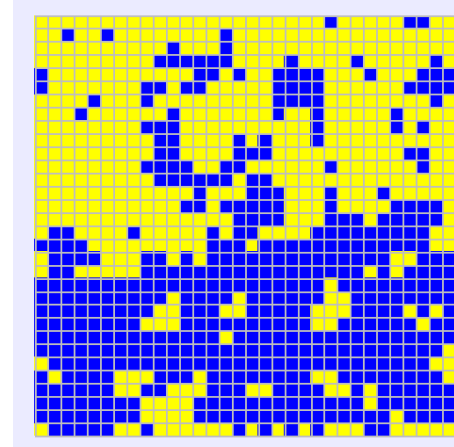
T=0.5



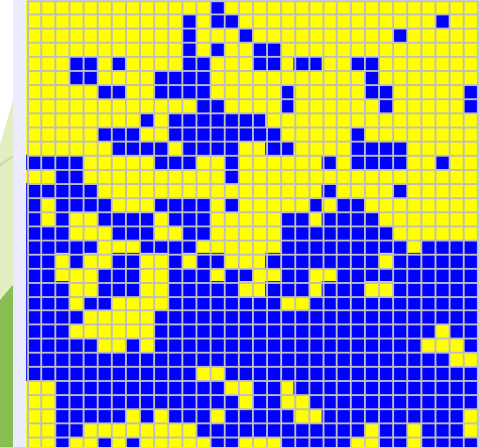
T=0.6



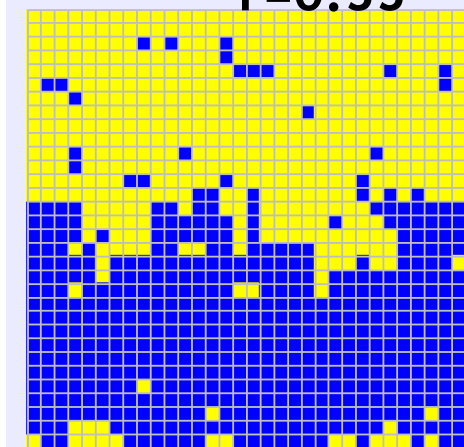
T=0.65



T=0.7



T=0.55



- (f) Simulate a lattice gas of $N = 2048$ particles on a $L = 64$ lattice at $T = 2.0$ with no gravitational field for 5000 mcs. Then change the temperature to $T = 0.2$. This process is called a (temperature) quench, and the resulting behavior is called *spinodal decomposition*. The domains grow very slowly as a function of time. Discuss why it is difficult for the system to reach its equilibrium state for which there is one domain of mostly occupied sites in equilibrium with one domain of mostly unoccupied sites. □

The screenshot displays a simulation environment on a Mac. The main window, 'LatGas2DApp Controller', has a menu bar (File, Edit, Display, Help) and an 'Input Parameters' table:

Name	Value
Length	64
Number of particles	2048
Temperature	2
Gravitational field	0
steps per display	100

Below the table are buttons for 'Start', 'Step', 'New', and 'Zero averages'. A 'Messages' section is at the bottom of the controller window.

On the left, the 'Particle Configuration' window shows a 64x64 grid with a horizontal split: the top half is yellow and the bottom half is blue. The 'Thermodynamic Quantities' window shows a plot of E/N (y-axis, ranging from -0.465 to -0.505) versus time (x-axis, ranging from 0.6 to 1.2 $\times 10^4$).

The background desktop features a sidebar with a list of simulation topics: Particle in a box, Second virial coefficient, Diffusive equilibrium, Chemical demon, Estimating the chemical potential, Potts model, XY model, FPU problem, Ideal quantum gases, Percolation, Renormalization group (percolation), **Lattice gas** (highlighted), and Diffusion on a lattice. The main desktop area contains text about Kawasaki dynamics and a section titled 'Problems' with a goal to observe two-phase coexistence and spinodal decomposition. The first problem states: '1. Run the program at $T = 0.4$ with 600 particles on a 32×32 lattice in zero gravitational field. You should see one region of particles (green sites) with a few small holes or bubbles and another region with just a few isolated particles or small clusters of particles. This system represents a liquid (the predominately green region) in equilibrium with its vapor (the mostly white region). Record the energy. To speed up the simulation set steps per display to 100.'