

Seminar 11

W2-1

Warmup Problems

Warmup
Problems

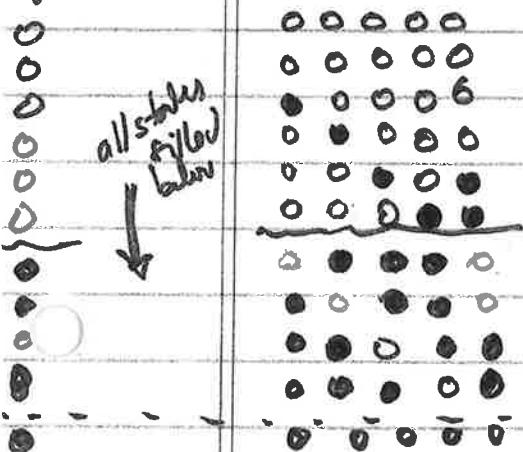
Counting Fermions

Schroeder 7.16

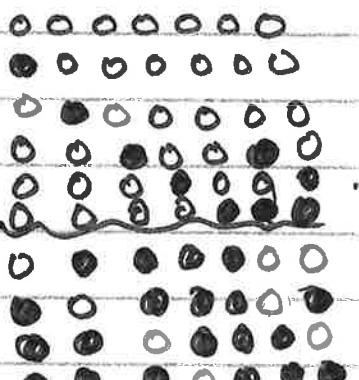
- (a) Draw "dot diagram" for $q=4, 5, 6$
(energy $4\gamma, 5\gamma, 6\gamma$) with N identical
Fermions ...

$q=4$

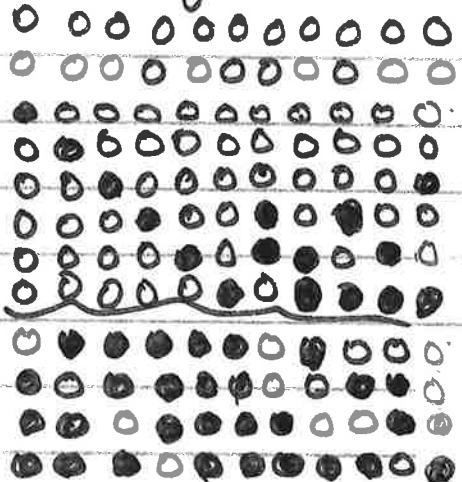
$q=0$



$q=5$



$q=6$



(b) $q=6 \rightarrow 11$ states on

$E = \frac{1}{h(E)} \text{ plot}$
 next page

level (cont) prob

level
"0" & This is called prob
 $E = \frac{1}{h(E)}$ for plot on next page

1	10	11
2	9	
3	8	
4	7	
5	6	
6	5	
7	4	
8	3	
9	2	

below
"9"
all states
are occupied
 $\rightarrow p_{ob} = 1/11$
for $q=6$

10	1/11
11	1/11
12	0
13	0
14	0

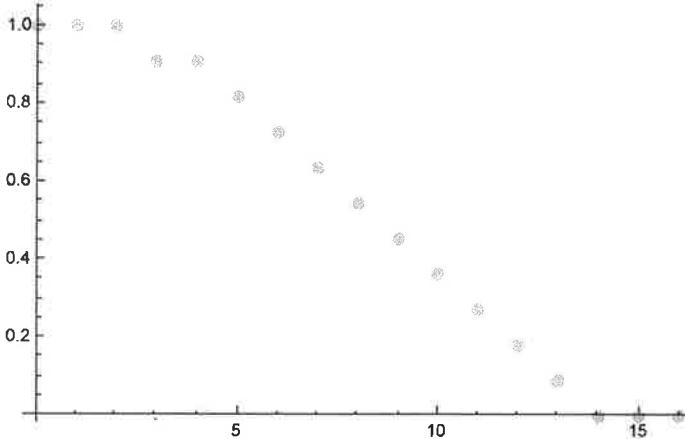
F_e is n(ϵ)

Here are occupancy numbers vs. energy for $q = 6$. We start with a few levels where all states are occupied so it looks more like a Fermi function! We similarly end with a bunch of unoccupied levels.

```
FermiList = {{0, 1}, {1, 1}, {2, 1}, {3, 0.90909}, {4, 0.90909}, {5, 0.818181},
{6, 0.727272}, {7, .636363}, {8, 0.545454}, {9, 0.454545}, {10, 0.36363},
{11, 0.272727}, {12, 0.181818}, {13, 0.090909}, {14, 0}, {15, 0}, {16, 0}}
```

```
 {{0, 1}, {1, 1}, {2, 1}, {3, 0.90909}, {4, 0.90909}, {5, 0.818181},
{6, 0.727272}, {7, 0.636363}, {8, 0.545454}, {9, 0.454545}, {10, 0.36363},
{11, 0.272727}, {12, 0.181818}, {13, 0.090909}, {14, 0}, {15, 0}, {16, 0}}
```

```
FermiPlot = ListPlot[FermiList]
```



```
ClearAll[x, T]
```

```
FindFit[Evaluate[FermiList], 1 / (Exp[(x - 8.5) / T] + 1), {T}, x]
```

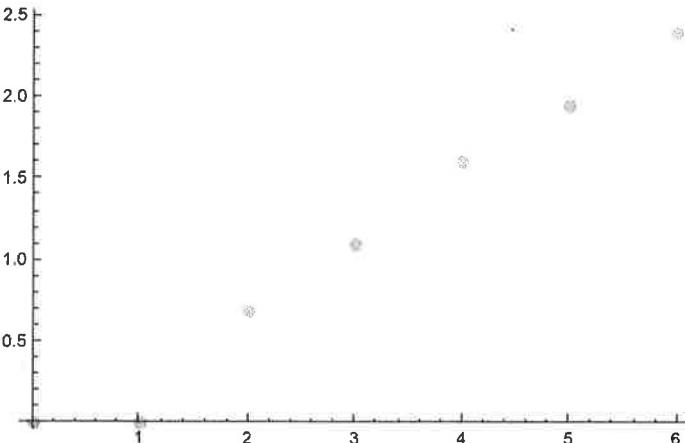
```
{T → 2.13806}
```

$$T \approx 2.14 \text{ eV/k}$$

```
EntropyList = {{0, 0}, {1, 0}, {2, 0.69}, {3, 1.10}, {4, 1.61}, {5, 1.95}, {6, 2.40}}
```

```
 {{0, 0}, {1, 0}, {2, 0.69}, {3, 1.1}, {4, 1.61}, {5, 1.95}, {6, 2.4}}
```

```
ListPlot[EntropyList]
```



(c) What is μ ?

It is where

$$\bar{n}(\epsilon) \approx \frac{1}{2} \text{ so}$$

I take it as

$$\mu = 8.5$$

Must keep in mind,

though my $\epsilon = 0$
is arbitrary.

See next
page for
data going
into EntropyList

```
In[1]: FindFit[Evaluate[EntropyList], a*x+b, {a, b}, x]
Out[1]: {a → 0.429286, b → -0.180714}
```

```
In[2]: Temperature = 1 / 0.429286
Out[2]: 2.32945
```

$$\frac{dS}{dE} = \frac{1}{T} \Rightarrow T = 2.33 \text{ K}$$

Not bad agreement with part (b)

(d) (cont)

<u>g</u>	<u>S_E</u>	<u>S/k = ln(S_E)</u>
0	1	0
1	1	0
2	2	0.69
3	3	1.10
4	5	1.61
5	7	1.95
6	11	2.40

from $g = 0, 1, 2, 3$
done in text